Tetrahedron 64 (2008) 5637–5644

Contents lists available at [ScienceDirect](www.sciencedirect.com/science/journal/00404020)

Tetrahedron

journal homepage: www.elsevier.com/locate/tet

On the influence of chiral auxiliaries in the stereoselective cross-coupling reactions of titanium enolates and acetals

Jessica Baiget, Annabel Cosp, Erik Gálvez, Loreto Gómez-Pinal, Pedro Romea*, Fèlix Urpí*

Departament de Química Orgànica, Universitat de Barcelona, Martí i Franqués 1-11, 08028 Barcelona, Catalonia, Spain

article info

Article history: Received 4 March 2008 Received in revised form 2 April 2008 Accepted 10 April 2008 Available online 15 April 2008

Keywords: Asymmetric synthesis Chiral auxiliaries Titanium enolates Acetals

1. Introduction

It is beyond all doubt that chiral auxiliaries have been greatly responsible for the amazing development of highly stereoselective methodologies occurred throughout the last decades, which have made possible the construction of structurally complex molecular architectures that were considered unattainable few years $ago¹$ Even in our days, when the state of the art in the asymmetric synthesis arena is currently associated to the accomplishments on catalysis, chiral auxiliary-based processes still maintain a prominent position among the most reliable strategies for accessing a single stereoisomer in high yields.² Moreover, a thorough and endless analysis of the influence of the structural features of many chiral auxiliaries on the stereochemical outcome of the processes in which they are involved has provided a better understanding of their reactivity and improved their synthetic efficiency.^{[2](#page-6-0)}

That is the case for oxazolidinones, probably the most representative family of chiral auxiliaries.³ Indeed, chiral 1,3-oxazolidinones, first introduced by Evans, 4 have turned out to be a superb tool for the stereoselective construction of carbon–carbon bonds and have enjoyed tremendous success. 5 Nevertheless, the diastereoselectivity imparted by these auxiliariesis occasionallylow, as for the acetate-like aldol reaction.^{4,6} Interestingly, Nagao and Fujita recognized that tin(II) enolate from analogous N-acetyl-1,3-thiazolidine-2-thiones could participate in highly stereoselective aldol processes.⁷ Further

ABSTRACT

Titanium enolates from chiral N-propanoyl-1,3-thiazolidine-2-thiones containing bulky substituents at C4 turned out to be excellent platforms to get highly stereocontrolled cross-coupling reactions with acetals. Related oxazolidinethiones also afforded good results, but the corresponding oxazolidinones resulted completely unselective for such reactions, which proves that an exocyclic C=S bond is essential to attain a synthetically useful stereocontrol.

- 2008 Elsevier Ltd. All rights reserved.

contributions have shown that other thiazolidinethiones also impart excellent levels of stereocontrol provided that a bulky substituent is incorporated to the heterocycle.^{[8](#page-6-0)} On the whole, it has been established that a sulfur atom at the exocyclic position offers different modes of metal binding for enolates and gives access to alternative stereoisomers. Thus, chiral 1,3-thiazolidine- and 1,3-oxazolidine-2-thiones represented in Figure 1 have emerged as suitable platforms to attain highly stereoselective carbon–carbon bond formation reactions.^{[7–10](#page-6-0)}

In this context, we have disclosed a straightforward entry to enantiopure *anti* β-alkoxy-α-methyl oxygenated derivatives based on the Lewis acid-mediated addition of titanium enolates from (S)-4-isopropyl-N-propanoyl-1,3-thiazolidine-2-thione to aliphatic, aromatic, and α , β -unsaturated acetals and the easy removal of the chiral auxiliary (Scheme 1).¹¹

Once established the synthetic potential of this new methodology, two structural issues remained still unclear: the influence of (i)

Figure 1.

^{*} Corresponding authors. Tel.: $+34$ 93 4021247/4039106; fax: $+34$ 93 3397878 $(F.U.)$

E-mail addresses: pedro.romea@ub.edu (P. Romea), felix.urpi@ub.edu (F. Urpı´).

^{0040-4020/\$ –} see front matter © 2008 Elsevier Ltd. All rights reserved. doi:10.1016/j.tet.2008.04.044

Scheme 2.

different 1,3-thiazolidine-2-thiones and (ii) related chiral auxiliaries on the stereochemical outcome of the process. Herein, we document the impact of different substituents R positioned at C4 in N-propanoyl-1,3-thiazolidine-2-thiones on the addition of their titanium enolates to a representative acetal, namely the dimethyl acetal of benzaldehyde (see Scheme 2). Additionally, these studies have been expanded to related 1,3-oxazolidine-2-thione and 1,3 oxazolidin-2-one chiral auxiliaries (see Scheme 2).

2. Results and discussion

At first, we were concerned with the stereochemical control imparted by the 1,3-thiazolidine-2-thiones 1 shown in Scheme 3. With the exception of **1f** (R: t-Bu), these sulfur-containing heterocycles were easily prepared by refluxing an alkaline solution (2.25 M KOH in 1:1 EtOH/H₂O) of the corresponding β -amino al-cohol and carbon disulfide for three days.^{[12](#page-6-0)} In turn, the thiazolidinethione 1f derived from the (S)-tert-leucinol required harsher conditions (5 M KOH in H_2O) due to the steric hindrance imposed by the bulky tert-butyl group (Scheme 3).^{8b} As a rule, such stringent conditions are needed to overcome the formation of the related 1,3 oxazolidine-2-thiones and to obtain the 1,3-thiazolidine-2-thiones 1 as a sole product, usually pure enough to be used in the next step without any chromatographic purification. Thus, such experimental procedure turns out to be particularly useful to prepare the chiral auxiliaries 1 in multigram scale. Eventually, their acylation with propanoyl chloride proceeded smoothly to afford the desired N-propanoyl-1,3-thiazolidine-2-thiones 2 in high yields.

With a reliable supply of a wide set of N-propanoyl-1,3-thiazolidine-2-thiones 2, we began to study the addition of their titanium enolates to the dimethyl acetal of benzaldehyde in the presence of BF_3 $-OEt_2$ under the conditions previously optimized.¹³ The results are summarized in Table 1.

Scheme 3. Reagents and conditions: (a) CS₂, KOH, EtOH/H₂O 1:1, reflux, 72 h. (b) EtCOCl, BuLi, THF, -78 °C. (c) CS₂, KOH, H₂O, reflux, 72 h.

Table 1

Determined by HPLC

^b Isolated yield of the 2,3-anti diastereomer 4; values in brackets denote overall yield.

Figure 2.

The absolute configuration of the 2,3-anti adduct 4e was firmly established through X-ray analysis (Fig. 2).^{[14](#page-6-0)} Other 2,3-anti relationships on major diastereomers 4 were secured by removal of the chiral auxiliaries with methanol and comparison of the resultant methyl esters **6** (Scheme 4). Furthermore, the $\frac{3}{2,3}$ coupling constants of **4** (${}^{3}J_{2,3}$ ${}^{3}J_{2,3}$ ${}^{3}J_{2,3}$ 9.8–9.9 Hz, see Section 3) and **5** (${}^{3}J_{2,3}$ 6.6– 7.0 Hz) resulted to be independent of the C4 substituent and were used as a diagnostic tool to assign the relative configuration of the adducts **4** and 5^{15} 5^{15} 5^{15}

Scheme 4. Reagents and conditions: (a) MeOH, K_2CO_3 , rt.

As shown in [Table 1,](#page-1-0) the diastereoselectivity achieved in such cross-coupling reactions depends on the heterocycles 1. The most simple N-propanoyl-1,3-thiazolidine-2-thione (R: H) 2a affords the 2,3-anti adduct 4a in a moderate selectivity (dr 65:35), which is improved by all the other C4 substituted chiral auxiliaries. In fact, the diastereoselectivity increases with steric bulk of R (compare entries 1 and 2–6 in [Table 1](#page-1-0))^{[16](#page-6-0)} to reach the best figures with 2e (R: *i*-Pr) and 2f (R: t-Bu), respectively (see entries 5 and 6 in [Table 1](#page-1-0)). From practical purposes, the valine-derived auxiliary 1e might be the most reasonable choice because the stereochemical control achieved by 2e and 2f is very close (dr 88:12 vs 90:10) being much better than the yield for $2e(87 \text{ vs } 57\%)$, see entries 5 and 6 in [Table 1\)](#page-1-0).^{[17,18](#page-7-0)}

Importantly, only two of four possible diastereomers, namely 2,3-anti 4 and 2,3-syn 5 adducts, are observed across all the reaction mixtures. Therefore, it occurs as the substituent positioned at C4 in 2 prevents the addition of the electrophile to the Re face of the enolate and is responsible for the R configuration of the α stereocenter in 4 and 5. This and other evidences suggest that the addition proceeds through a S_N1 -like mechanism that involves the formation of an oxonium cation.^{[19–21](#page-7-0)} Then, such oxonium cation reacts with the less sterically hindered face of a chelated enolate following an antiperiplanar approach in an open transition state (Scheme 5). This mechanistic picture would be common for all the N-propanoyl-1,3-thiazolidine-2-thiones 2. Regarding the influence of R on the configuration of the β -stereocenter, we are aware that this model does not account easily for the differences on diastereoselectivity. We just speculate that the improvement of diastereomeric ratios achieved from achiral 2a to the more bulky 2f might be due to subtle changes on the structure of the enolate produced by the substituent positioned at $C4²²$ $C4²²$ $C4²²$

Having disclosed the influence of 2 on the stereochemical outcome of the Lewis acid-mediated addition of their titanium enolates to $PhCH(OMe)_2$, we next evaluated the behavior of N-propanoyl 1,3-oxazolidine-2-thiones 7 and 1,3-oxazolidin-2-one 8e containing isopropyl and tert-butyl groups at C4 (Fig. 3).

The N-propanoyl 1,3-oxazolidin-2-thiones 7 were prepared in good yields by acylation with propanoyl chloride of the chiral 1,3-oxazolidin-2-thiones 3, which, in turn, were obtained from the corresponding amino alcohols (Scheme 6). As represented in [Scheme 3](#page-1-0), (S)-4-tert-butyl-1,3-oxazolidine-2-thione (3f) was unexpectedly obtained when (S)-tert-leucinol was treated with carbon disulfide in a strong basic solution for a long time. Conversely, the less bulky (S) -4-isopropyl-1,3-oxazolidine-2-thione (3e) required milder experimental conditions and was isolated in good yield by simple stirring at room temperature in the presence of a weak base as triethylamine. Otherwise, standard acylation of the commercially available (S)-4-isopropyl-1,3-oxazolidin-2-one with propanoyl chloride delivered the desired (S)-4-isopropyl-N-propanoyl-1,3-oxazolidin-2-one (8e) in 81% yield.

Scheme 6. Reagents and conditions: (a) CS_2 , Et_3N , THF, reflux, 12 h [for R: t-Bu, see [Scheme 3\]](#page-1-0). (b) EtCOCl, BuLi, THF, -78 °C.

Scheme 5.

Unfortunately, the N-acyl oxazolidinone 8e proved to be stereochemically unselective in the cross-coupling reaction with the dimethyl acetal of benzaldehyde. Indeed, the addition of the titanium enolate from **8e** to PhCH(OMe) $_2$ in the presence of a wide array of Lewis acids ($BF_3 \cdot OEt_2$, BCl_3 , $SnCl_4$, $TiCl_4$, etc.) proceeded smoothly and delivered the corresponding adducts in good yields but in unacceptable low diastereomeric ratios (dr<60:40). Interestingly, only two diastereomers were observed, which proves the absolute control exerted by the isopropyl group positioned at C4 on the configuration of the new α -stereocenter.

Better results were achieved with the N-acyl oxazolidinethiones 7 [\(Fig. 3\)](#page-2-0). As shown in Scheme 7, both (S)-4-isopropyl-N-propanoyl-1,3-oxazolidine-2-thione (7e) and (S)-4-tert-butyl-N-propanoyl-1,3-oxazolidine-2-thione (7f) afforded the corresponding adducts in close diastereoselectivity than the related thiazolidinethiones 2e and 2f, which proves that the stereochemical control on these cross-coupling reactions is associated to the presence of an exocyclic sulfur atom $(C=S)$.^{[23](#page-7-0)} Once again, the most sterically hindered tert-leucine derived oxazolidinethione 7f afforded the 2,3-anti adduct 9f in higher diastereomeric ratio but in modest yield.

Scheme 7. Reagents and conditions: (a) (i) TiCl₄, i -Pr₂NEt, CH₂Cl₂. (ii) BF₃ OEt₂, PhCH(OMe)₂.

In conclusion, the stereochemical outcome of the cross-coupling reactions of titanium enolates from C4 substituted N-propanoyl-1,3thiazolidine-2-thiones with $PhCH(OMe)_2$ in the presence of $BF_3 \cdot OEt_2$ relies on the steric bulk of the C4 substituent, being the valine-derived chiral auxiliary the best choice to obtain the corresponding 2,3-anti adduct in high yield and diastereomeric ratio. Related 1,3-oxazolidine-2-thiones afford similar results but in poorer yield and diastereoselectivity. Furthermore, the valine-derived 1,3-oxazolidin-2-one resulted useless for such reactions. Thus, the presence of a sulfur atom in the exocyclic double bond is essential to attain a synthetically useful stereocontrol.

3. Experimental

3.1. General

Melting points were taken on an Electrothermal apparatus and are uncorrected. Specific rotations were determined at 20° C on a Perkin–Elmer 241 MC polarimeter. IR spectra were recorded on a Nicolet 510FT spectrometer and only the more representative frequencies (cm $^{-1}$) are reported. ¹H NMR (300 MHz) and ¹³C NMR (75.4 MHz) spectra were recorded on a Varian Unity Plus spectrometer; ¹H NMR (400 MHz) spectra were recorded on a Varian Mercury; chemical shifts (δ) are quoted in parts per million and referenced to internal TMS for ¹H NMR and CDCl₃ (δ 77.0) for ¹³C NMR; data are reported as follows: s, singlet; d, doublet; t, triplet; q, quartet; hept, heptuplet; oct, octet; m, multiplet; br, broad; coupling constants (J) are quoted in hertz; where appropriate, 2D techniques were also used to assist in structural elucidation. Mass spectra were obtained from the Centro de Apoio Cientifico Tecnoloxico a Investigacion (C.A.C.T.I.), Universidad de Vigo and from the Servei d'Espectrometria de Masses, Universitat de Barcelona. Elemental analyses were obtained from the Servei de Microanàlisi (CID–CSIC, Barcelona). HPLC was carried out with a silica gel Spherisorb S3W (250 \times 4 mm) column with a 0.9 mL min⁻¹ flux. Flash chromatography was performed on SDS silica gel $(35-70 \,\mu m)$. Analytical thin layer chromatography was carried out on Merck Kieselgel 60 $F₂₅₄$ plates. The following solvents and reagents were purified and dried according to the standard procedures: $CH₂Cl₂$, THF, and i -Pr₂NEt. All other reagents were used as received.

3.2. General procedure for the preparation of 1,3 thiazolidine-2-thiones (1)

A 2.25 M KOH in 1:1 EtOH/H2O solution (120 mL) was added dropwise to a solution of enantiomerically pure β -amino alcohol (0.1 mol) and carbon disulfide (18 mL, 0.265 mol) in ethanol (30 mL) at room temperature and under N_2 . The reaction mixture was stirred and heated at reflux for 72 h under N_2 . After cooling, the volatiles were removed with a rotary evaporator. Then, a 0.5 M aqueous HCl (350 mL) was carefully added at room temperature and the resulting mixture was extracted with CH_2Cl_2 (3×100 mL). The combined organic layers were dried (MgSO4), filtered, and the solvent was removed to give the corresponding 1,3-thiazolidine-2-thiones 1 as a solid, which was usually used in the next step without further purification. Otherwise, this material was purified through a short column chromatography to afford purer samples.

3.2.1. 1,3-Thiazolidine-2-thione (1a)

Yield: 58%; white solid; R_f 0.20 (CH₂Cl₂); mp 105-106 °C; IR (KBr): 3141, 2849, 1516, 1296, 1051; ¹H NMR (CDCl₃, 300 MHz): δ 8.00 (1H, br s, NH), 4.04-3.96 (2H, AA'BB'X system, NCH₂), 3.61-3.54 (2H, AA'BB'X system, SCH₂); ¹³C NMR (CDCl₃, 75.4 MHz): δ 201.8, 51.3, 33.6; MS–CI (NH₃): m/z (%) 120 [M+H]⁺ (100).

3.2.2. (S)-4-Phenyl-1,3-thiazolidine-2-thione $(1b)$

Yield: 77%; white solid; R_f 0.40 (CH₂Cl₂); mp 124-126 °C [lit.^{12a} mp 124–125 °C]; $[\alpha]_D$ –213.5 (c 1.1, CHCl₃) [lit.^{12a} $[\alpha]_D$ –209.3 (c 0.35, CHCl3)]; IR (KBr): 3124, 2950, 1492, 1453, 1258, 1050; ¹H NMR $(CDC1_3, 300 MHz)$: δ 7.60 (1H, br s, NH), 7.50–7.40 (5H, m, ArH), 5.35–5.25 (1H, m, NCH), 3.85 (1H, dd, $J=11.2$, 8.0, SCH_xH_y), 3.51 (1H, dd, J=11.2, 8.3, SCH_xH_y); ¹³C NMR (CDCl₃, 75.4 MHz): δ 201.7, 138.0, 129.3, 129.2, 126.2, 67.3, 41.6; MS–CI (NH₃): m/z (%) 196 $[M+H]$ ⁺ (100).

3.2.3. (S)-4-Isobutyl-1,3-thiazolidine-2-thione $(1c)$

Yield: 50%; white solid; R_f 0.40 (CH₂Cl₂); mp 48-50 °C; [α]_D -28.8 (c 1.0, CHCl3); IR (KBr): 3143, 2957, 2869, 1507, 1467, 1305, 1033; ¹H NMR (CDCl₃, 300 MHz): δ 8.06 (1H, br s, NH), 4.38-4.26 (1H, m, NCH), 3.59 (1H, dd, J=11.0, 7.7, SCH_xH_y), 3.22 (1H, dd, J=11.0, 7.9, SCH_xH_y), 1.80-1.60 (2H, m, CH_xH_yCH(CH₃)₂), 1.55-1.45 (1H, m, $CH_XH_VCH(CH_3)_2$, 0.97 (3H, d, J=6.6, CH₃), 0.96 (3H, d, J=6.4, CH₃); ¹³C NMR (CDCl₃, 75.4 MHz): δ 200.7, 62.5, 42.9, 38.9, 25.3, 22.6, 22.2. Anal. Calcd for C₇H₁₃NS₂: C, 47.96; H, 7.47; N, 7.99. Found: C, 47.91; H, 7.50; N, 7.71.

3.2.4. (S)-4-Benzyl-1,3-thiazolidine-2-thione (1d)

Yield: 95%; white solid; R_f 0.35 (CH₂Cl₂); mp 74–75 °C [lit.^{[12a](#page-6-0)} mp 84–85 °C]; $[\alpha]_{\text{D}}$ –119.2 (c 1.0, CHCl₃) [lit.^{[12a](#page-6-0)} $[\alpha]_{\text{D}}$ –129.2 (c 0.96, CHCl₃)]; IR (KBr): 3134, 2969, 1509, 1494, 1004; ¹H NMR (CDCl₃, 300 MHz): d 7.45 (1H, br s, NH), 7.40–7.26 (3H, m, ArH), 7.23–7.17 $(2H, m, ArH), 4.52-4.40$ (1H, m, NCH), 3.60 (1H, dd, J=11.2, 7.7, SCH_xH_y), 3.33 (1H, dd, J=11.2, 6.8, SCH_xH_y), 3.03 (1H, dd, J=13.6, 7.6, PhCH_aH_b), 2.98 (1H, dd, J=13.6, 6.7, PhCH_aH_b); ¹³C NMR (CDCl₃, 75.4 MHz): δ 200.8, 135.7, 129.0 (×2), 127.4, 65.0, 39.9, 38.0; MS-CI (NH_3) : m/z (%) 210 $[M+H]$ ⁺ (100).

3.2.5. (S)-4-Isopropyl-1,3-thiazolidine-2-thione (1e)

Yield: 92%; white solid; R_f 0.25 (CH₂Cl₂); mp 68–69 °C [lit.^{[7](#page-6-0)} mp 67–68 °C, lit.^{[12a](#page-6-0)} mp 66–6[7](#page-6-0) °C]; [α]_D –34.9 (c 1.1, CHCl₃) [lit.⁷ [α]_D

 -36.8 (c 1.16, CHCl₃), lit.^{[12a](#page-6-0)} [α]_D -34.6 (c 0.94, CHCl₃)]; IR (KBr): 3190, 2965, 1500, 1410, 1385, 1030; ¹H NMR (CDCl₃, 300 MHz): δ 7.42 (1H, br s, NH), 4.03 (1H, td, J=8.2, 7.8, NCH), 3.52 (1H, dd, J=11.0, 8.2, SCH_xH_y), 3.34 (1H, dd, J=11.0, 8.2, SCH_xH_y), 2.10-1.95 (1H, m, CH(CH₃)₂), 1.04 (3H, d, J=7.0, CH₃), 1.01 (3H, d, J=7.2, CH₃); ¹³C NMR (CDCl₃, 75.4 MHz): δ 201.5, 70.6, 36.5, 32.5, 19.3, 18.7. Anal. Calcd for $C_6H_{11}NS_2$: C, 44.68; H, 6.87; N, 8.68. Found: C, 44.74; H, 6.81; N, 8.55.

3.3. Preparation of (S)-4-tert-butyl-1,3-thiazolidine-2 thione (1f)

A mixture of (S)-tert-leucinol (2.09 g, 17.8 mmol), carbon disulfide (9.2 mL, 150 mmol) in 5 M KOH aqueous solution (100 mL) was heated for 72 h under N_2 . The reaction mixture was extracted with CH₂Cl₂ (3 \times 100 mL), the organic layers were dried (MgSO₄), and concentrated. The resulting oil was purified by flash chromatography (CH_2Cl_2) to afford 1.74 g (9.9 mmol, 55%) of (S)-4-tert-butyl-1,3-thiazolidine-2-thione (1f). White solid; R_f 0.65 (CH₂Cl₂); mp 134–135 °C [lit.^{[24](#page-7-0)} mp 143–144 °C]; [α]_D –35.8 (c 1.1, CHCl₃) [lit.²⁴ $[\alpha]_{\text{D}}$ –33.2 (c 1.03, CHCl₃)]; IR (KBr): 3157, 2958, 1507, 1475, 1368, 1293, 1042; ¹H NMR (CDCl₃, 300 MHz): δ 7.60 (1H, br s, NH), 4.07-3.98 (1H, m, NCH), 3.45 (1H, dd, J=11.3, 8.4, SCH_xH_v), 3.39 (1H, dd, J=11.3, 9.0, SCH_xH_y), 1.02 (9H, s, C(CH₃)₃); ¹³C NMR (CDCl₃, 75.4 MHz): δ 201.6, 73.4, 34.4 (\times 2), 25.9; HRMS (+ESI): calcd for $[M+H]$ ⁺ C₇H₁₄NS₂ 176.0562, found 176.0567.

3.4. General procedure for the acylation of the 1,3 thiazolidine-2-thiones

A 1.5 M solution of n-BuLi in hexanes (7.4 mL, 11 mmol) was added dropwise to a solution of 1,3-thiazolidine-2-thione 1 (10 mmol) in THF (6.6 mL) at -78 °C under N₂. The reaction mixture was stirred for 15 min and propanoyl chloride (1.1 mL, 12.5 mmol) was carefully added. The resulting clear solution was stirred for 5 min and the solution was allowed to warm to room temperature and stirred for 1.5 h. The reaction mixture was cooled with an ice-water bath and was quenched with a saturated solution of NH4Cl (4 mL) and water (10 mL). This mixture was extracted with CH_2Cl_2 (3×20 mL), the combined organic layers were dried (MgSO4), filtered, and concentrated. The resultant oil was purified through column chromatography to afford the corresponding Npropanoyl-1,3-thiazolidine-2-thione 2.

3.4.1. N-Propanoyl-1,3-thiazolidine-2-thione (2a)

Yield: 58%; bright yellow oil; R_f 0.65 (CH₂Cl₂); IR (film): 2979, 2938, 1702, 1365, 1281, 1222, 1157; ¹H NMR (CDCl₃, 300 MHz): δ 4.59 (2H, t, J=7.6, NCH₂), 3.28 (2H, t, J=7.6, SCH₂), 3.26 (2H, q, J=7.3, COCH₂), 1.18 (3H, t, J=7.3, CH₃); ¹³C NMR (CDCl₃, 75.4 MHz): d 201.5, 175.5, 56.0, 32.2, 28.2, 8.7; MS–CI (NH3): m/z (%) 176 $[M+H]^{+}$ (100).

3.4.2. (S)-4-Phenyl-N-propanoyl-1,3-thiazolidine-2-thione (2b)

Yield: 84%; viscous yellow oil; $R_{\it f}$ 0.65 (CH₂Cl₂); [$\alpha|_{\rm D}$ –301.3 (c 1.0, CHCl₃); IR (KBr): 2977, 1692, 1320, 1254, 1233, 1164, 1052; ¹H NMR (CDCl₃, 300 MHz): δ 7.40–7.30 (5H, m, ArH), 6.24 (1H, dd, J=8.2, 1.6, NCH), 3.93 (1H, dd, J=11.3, 8.2, SCH_xH_v), 3.38 (1H, dq, J=18.1, 7.3, COCH_xH_y), 3.20 (1H, dq, J=18.1, 7.3, COCH_xH_y), 3.07 (1H, dd, J=11.3, 1.6, SCH_xH_y), 1.13 (3H, t, J=7.3, COCH₂CH₃); ¹³C NMR (CDCl3, 75.4 MHz): d 202.1, 174.8, 139.3, 129.0, 128.4, 125.4, 69.8, 36.6, 32.5, 8.7; HRMS (+ESI): calcd for $[M+H]^{+}$ C₁₂H₁₄NOS₂ 252.0511, found 252.0509.

3.4.3. (S)-4-Isobutyl-N-propanoyl-1,3-thiazolidine-2-thione (2c)

Yield: 81%; bright yellow oil; R_f 0.45 (50:50 CH₂Cl₂/hexanes); $[\alpha]_D$ +269.8 (c 1.05, CHCl₃); IR (film): 2959, 1701, 1341, 1269, 1161,

1099; ¹H NMR (CDCl₃, 300 MHz): δ 5.35-5.20 (1H, m, NCH), 3.56 (1H, ddd, J=11.2, 7.2, 1.2, SCH_xH_v), 3.36 (1H, dq, J=18.1, 7.2, COCH_xH_y), 3.10 (1H, dq, J=18.1, 7.2, COCH_xH_y), 2.91 (1H, dd, J=11.2, 0.7, SCH_xH_y), 1.98-1.86 (1H, m, CH_xH_yCH(CH₃)₂), 1.74-1.48 (2H, m, CH_xH_yCH(CH₃)₂), 1.17 (3H, t, J=7.2, COCH₂CH₃), 1.01 (3H, d, J=6.4, CHCH₃), 1.00 (3H, d, J=6.4, CHCH₃); ¹³C NMR (CDCl₃, 75.4 MHz): d 201.4, 174.7, 66.1, 39.6, 33.0, 32.2, 25.4, 23.5, 21.2, 8.8; HRMS (+ESI): calcd for $[M+H]^+$ C₁₀H₁₈NOS₂ 232.0824, found 232.0828.

3.4.4. (S)-4-Benzyl-N-propanoyl-1,3-thiazolidine-2-thione (2d)

Yield: 89%; yellow solid; R_f 0.40 (50:50 CH₂Cl₂/hexanes); mp 95–97 °C; α _D + 196.4 (c 1.1, CHCl₃); IR (KBr): 2977, 1706, 1264, 1165, 1032; ¹H NMR (CDCl₃, 400 MHz): δ 7.38-7.26 (5H, m, ArH), 5.42-5.35 (1H, m, NCH), 3.42 (1H, dq, J=18.2, 7.2, COCH_xH_v), 3.38 (1H, ddd, J = 11.5, 7.2, 1.2, SCH_xH_v), 3.22 (1H, dd, J = 13.1, 3.8, PhCH_xH_v), 3.13 (1H, dq, J=18.2, 7.2, COCH_xH_v), 3.05 (1H, dd, J=13.1, 10.5, PhCH_xH_y), 2.88 (1H, dd, J=11.5, 0.7, SCH_xH_y), 1.19 (3H, t, J=7.2, CH₃); ¹³C NMR (CDCl₃, 75.4 MHz): δ 201.1, 174.9, 136.6, 129.4, 128.9, 127.2, 68.6, 36.7, 32.3, 31.9, 8.8. Anal. Calcd for C₁₃H₁₅NOS₂: C, 58.83; H, 5.70; N, 5.28. Found: C, 58.67; H, 5.62; N, 5.22.

3.4.5. (S)-4-Isopropyl-N-propanoyl-1,3-thiazolidine-2-thione (2e)

Yield: 98%; bright yellow oil; R_f 0.45 (50:50 CH₂Cl₂/hexanes); $[\alpha]_{\text{D}}$ +428.5 (c 1.0, CHCl₃); IR (film): 2961, 1699, 1349, 1259, 1167; ¹H NMR (CDCl₃, 300 MHz): δ 5.17 (1H, ddd, J=8.1, 6.1, 1.2, NCH), 3.51 (1H, dd, J=11.4, 8.1, SCH_xH_y), 3.36 (1H, dq, J=18.0, 7.2, COCH_xH_y), 3.15 (1H, dq, J=18.0, 7.2, COCH_xH_v), 3.02 (1H, dd, J=11.4, 1.2, SCH_xH_y), 2.41–2.32 (1H, m, CH(CH₃)₂), 1.15 (3H, t, J=7.2, COCH₂CH₃), 1.06 (3H, d, J=6.6, CH₃CCH₃), 0.98 (3H, d, J=6.6, CH₃CCH₃); ¹³C NMR (CDCl3, 75.4 MHz): d 203.3, 174.8, 71.6, 32.0, 30.8, 30.4, 19.0, 17.6, 8.9. Anal. Calcd for C₉H₁₅NOS₂: C, 49.73; H, 6.96; N, 6.44. Found: C, 49.67; H, 6.94; N, 6.36.

3.4.6. (S)-4-tert-Butyl-N-propanoyl-1,3-thiazolidine-2-thione (2f)

Yield: 93%; yellow solid; R_f 0.50 (50:50 CH₂Cl₂/hexanes); mp 29–30 °C; α _D +602.4 (c 0.85, CHCl₃); IR (KBr): 2964, 1703, 1353, 1325, 1250, 1156, 1045; ¹H NMR (CDCl₃, 300 MHz): δ 5.35 (1H, dd, $J=8.4$, 0.7, NCH), 3.54 (1H, dd, J=11.4, 8.4, SCH_xH_v), 3.36 (1H, dq, J=17.9, 7.2, COCH_xH_v), 3.21 (1H, dq, J=17.9, 7.2, COCH_xH_v), 3.10 (1H, dd, J=11.4, 0.7, SCH_xH_v), 1.18 (3H, t, J=7.2, COCH₂CH₃), 1.03 (9H, s, $C(CH_3)_3$; ¹³C NMR (CDCl₃, 75.4 MHz): δ 204.7, 174.5, 72.4, 37.9, 31.8, 30.5, 26.8, 9.2.

3.5. General procedure for the titanium-mediated addition of 2 to PhCH(OMe)₂

Neat TiCl4 (0.12 mL, 1.1 mmol) was added dropwise to a solution of 2 (1.0 mmol) in anhydrous CH₂Cl₂ (8 mL), at 0 °C under N₂. The yellow suspension was stirred for 5 min at 0 °C, cooled at -78 °C, and a solution of anhydrous diisopropylethylamine (0.19 mL, 1.1 mmol) in anhydrous CH_2Cl_2 (1 mL) was added. The dark red enolate solution was stirred for 2 h at -40 °C and cooled at -78 °C. Then, $BF_3 \cdot OEt_2$ (140 µL, 1.1 mmol) and PhCH(OMe)₂ (166 µL, 1.1 mmol) were successively added dropwise. The resulting mixture was stirred at -78 °C for 2.5 h. The reaction was quenched by the addition of a saturated solution of $NH₄Cl$ (6 mL) with vigorous stirring and the layers were separated. The aqueous layer was reextracted with CH_2Cl_2 (10 mL), and the combined organic extracts were washed with brine (10 mL), dried (Na₂SO₄), filtered, and concentrated. The residue was analyzed by HPLC and purified through flash column chromatography with deactivated silica gel $(2.5\%$ Et₃N) to afford the 2,3-anti adduct 4 as a pure diastereomer or as a mixture with the 2,3-syn adduct 5. Eventually, the product was kept in the fridge under nitrogen atmosphere to avoid undesired decompositions.

3.5.1. (\pm) N-[2,3-anti-3-Methoxy-2-methyl-3-phenylpropanoyl]-1,3-thiazolidine-2-thione $(4a)$

Yield: 40%; yellow solid; R_f 0.70 (CH₂Cl₂); mp 132-133 °C; IR (KBr): 2981, 2934, 1700, 1364, 1278, 1153; ¹H NMR (CDCl₃, 300 MHz): δ 7.42–7.28 (5H, m, ArH), 4.72–4.60 (2H, m, COCHCH₃, NCH_xH_y), 4.46 (1H, ddd, J=11.9, 11.1, 7.6, NCH_xH_y), 4.33 (1H, d, J=9.8, CHOCH₃), 3.43 (1H, td, J=11.1, 7.8, SCH_xH_v), 3.21 (1H, ddd, J=11.1, 7.6, 3.2, SCH_xH_v), 3.11 (3H, s, OCH₃), 0.95 (3H, d, J=6.7, COCHCH₃); ¹³C NMR (CDCl₃, 75.4 MHz): d 201.6, 178.6, 138.8, 128.5, 128.3, 127.9, 87.6, 56.8, 56.5, 46.2, 28.9, 14.1; MS–CI (NH₃): m/z (%) 264 [M $-$ OMe] $^+$ (60), 296 $[M+H]^{+}$ (100).

3.5.2. (S)-N-[(2R,3S)-3-Methoxy-2-methyl-3-phenylpropanoyl]-4 phenyl-1,3-thiazolidine-2-thione (4b)

Yield: 64%; bright yellow oil; $R_{\it f}$ 0.55 (CH $_2$ Cl $_2$); [$\alpha]_{\rm D}$ – 175.0 (c 1.0, CHCl₃); HPLC (98:2 hexanes/EtOAc) t_R 54.6 min [2,3-syn diastereomer, t_R 38.0 min]; IR (film): 2931, 1702, 1457, 1302, 1254, 1150, 1025; ¹H NMR (CDCl₃, 300 MHz): δ 7.50–7.20 (10H, m, ArH), 6.34 (1H, dd, J=8.2, 3.3, NCH), 5.08 (1H, dq, J=9.8, 6.9, COCHCH₃), 4.19 (1H, d, J=9.8, CHOCH₃), 3.84 (1H, dd, J=11.2, 8.2, SCH_xH_v), 3.14 (1H, dd, J=11.2, 3.3, SCH_xH_v), 2.87 (3H, s, OCH₃), 0.85 (3H, d, J=6.9, COCHCH₃); ¹³C NMR (CDCl₃, 75.4 MHz): δ 202.3, 177.3, 139.0, 138.9, 128.7, 128.4, 128.2 (2), 128.0, 126.0, 87.7, 70.3, 56.5, 45.5, 36.2, 14.4.

3.5.3. (S)-4-Isobutyl-N-[(2R,3S)-3-methoxy-2-methyl-3 phenylpropanoyl]-1,3-thiazolidine-2-thione $(4c)$

Yield of the mixture: 88%; yellow viscous oil; R_f 0.60 (CH₂Cl₂); HPLC (97:3 hexanes/EtOAc) t_R 13.4 min [2,3-syn diastereomer, t_R 14.5 min]; ¹H NMR (CDCl₃, 300 MHz): δ 7.40-7.20 (5H, m, ArH), 5.45-5.35 (1H, m, NCH), 5.08 (1H, dq, $J=9.9$, 7.0, COCHCH₃), 4.36 (1H, d, J=9.9, CHOCH₃), 3.53 (1H, ddd, J=11.1, 7.6, 0.7, SCH_xH_v), 3.10 (3H, s, OCH₃), 2.93 (1H, dd, J=11.1, 2.6, SCH_xH_v), 2.00–1.60 (3H, m, $CH_2CH(CH_3)_2$, 1.02 (3H, d, J=6.2, CH₃), 1.01 (3H, d, J=6.4, CH₃), 0.88 (3H, d, $J = 7.0$, CH₃); ¹³C NMR (CDCl₃, 75.4 MHz): δ 201.5, 177.9, 139.0, 128.4,128.2,128.0, 87.7, 66.5, 56.5, 45.5, 40.7, 33.0, 25.3, 23.6, 21.3,14.4.

3.5.4. (S)-4-Benzyl-N-[(2R,3S)-3-methoxy-2-methyl-3 phenylpropanoyl]-1,3-thiazolidine-2-thione (4d)

Yield of the mixture: 73%. A small amount of pure 4d was isolated and characterized. Bright yellow oil; R_f 0.17 (70:30 hexanes/CH₂Cl₂); HPLC (99:1 hexanes/EtOAc) t_R 36.6 min [2,3-syn diastereomer, t_R 46.2 min]; [a]D -41.2 (c 1.05 in CHCl3); IR (film): 2932, 1697, 1454, 1344, 1252, 1160, 1097; ¹H NMR (CDCl₃, 300 MHz): δ 7.40-7.25 (10H, m, ArH), 5.60–5.45 (1H, m, NCH), 5.22 (1H, dq, J=9.8, 6.9, COCHCH₃), 4.42 (1H, d, J=9.8, CHOCH₃), 3.40–3.25 (2H, m, PhCH_xH_y and SCH_XH_Y), 3.12 (3H, s, OCH₃), 3.08 (1H, dd, J=13.4, 10.5, PhCH_xH_y), 2.90 (1H, dd, J=11.4, 2.0, SCH_xH_y), 0.91 (3H, d, J=6.9, COCHCH₃); ¹³C NMR (CDCl3, 75.4 MHz): d 201.4, 177.8, 139.0, 136.8, 129.4, 128.9, 128.5, 128.3, 128.1, 127.2, 87.6, 68.9, 56.7, 45.4, 37.4, 31.5, 14.5; HRMS (+ESI): calcd for $[M-OMe]^+$ C₂₀H₂₀NOS₂ 354.0980, found 354.0964; calcd for $[M+H]^+$ C₂₁H₂₄NO₂S₂ 386.1242, found 386.1233.

3.5.5. (S)-4-Isopropyl-N-[(2R,3S)-3-methoxy-2-methyl-3 phenylpropanoyl]-1,3-thiazolidine-2-thione (4e)

Yield: 75%; yellow solid; mp 82–83 °C; R_f 0.60 (CH₂Cl₂); HPLC (97:3 hexanes/EtOAc) t_R 17.1 min [2,3-syn diastereomer, t_R 38.0 min]; α _D +120.2 (c 1.96 in CHCl₃); IR (KBr): 2927, 1704, 1368, 1245, 1156; ¹H NMR (CDCl₃, 300 MHz): δ 7.36-7.22 (5H, m, ArH), 5.36 (1H, ddd, J=8.9, 5.5, 2.1, NCH), 5.26 (1H, dq, J=9.8, 7.0, COCHCH₃), 4.35 (1H, d, J=9.8, CHOCH₃), 3.46 (1H, dd, J=11.5, 8.9, SCH_xH_v), 3.08 (3H, s, OCH₃), 3.01 (1H, dd, J=11.5, 2.1, SCH_xH_v), 2.40 $(1H, \text{heptd}, J=8.0, 5.5, \text{CH}(\text{CH}_3)_2)$, 1.11 (3H, d, J = 8.0, CH₃CHCH₃), 1.03 (3H, d, J=8.0, CH₃CHCH₃), 0.86 (3H, d, J=7.0, COCHCH₃); ¹³C NMR (CDCl3, 75.4 MHz): d 202.6, 177.7, 139.1, 128.3, 128.2, 128.1, 87.7, 71.9, 56.4, 45.1, 30.4, 28.8, 19.0, 17.0, 14.3; HRMS (+FAB): calcd for $[M+H]$ ⁺ C₁₇H₂₄NO₂S₂ 338.1249, found 338.1251.

3.5.6. (S)-4-tert-Butyl-N-[(2R,3S)-3-methoxy-2-methyl-3 phenylpropanoyl]-1,3-thiazolidine-2-thione (4f)

Yield of the mixture: 57%; bright yellow oil; R_f 0.40 (95:5 hexanes/EtOAc); HPLC (99:1 hexanes/EtOAc) t_R 41.4 min [2,3-syn diastereomer, t_R 37.1 min]; IR (film): 2965, 1698, 1454, 1370, 1242, 1134; ¹H NMR (CDCl₃, 400 MHz): δ 7.38-7.29 (5H, m, ArH), 5.53 (1H, dd, J=8.5, 0.9, NCH), 5.42 (1H, dq, J=10.0, 6.9, COCHCH₃), 4.37 (1H, d, J = 10.0, CHOCH₃), 3.49 (1H, dd, J = 11.7, 8.5, SCH_xH_v), 3.09 (3H, s, OCH₃), 3.07 (1H, dd, J=11.7, 0.9, SCH_xH_y), 1.10 (9H, s, C(CH₃)₃), 0.80 $(3H, d, J=6.9, COCHCH₃);$ 13C NMR (CDCl₃, 75.4 MHz): δ 205.0, 176.3, 139.3, 128.4, 128.2, 128.1, 87.2, 72.5, 56.5, 44.1, 37.9, 29.5, 26.7, 14.3; HRMS (+ESI): calcd for $[M-OMe]^+$ C₁₇H₂₂NOS₂ 320.1137, found 320.1149; calcd for $[M+H]^+$ C₁₈H₂₆NO₂S₂ 352.1399, found 352.1408.

3.6. General procedure for the removal of chiral auxiliaries: obtention of methyl ester 6

A solution of 4 (0.15 mmol) in anhydrous MeOH (3 mL) was treated with anhydrous K_2CO_3 (104 mg, 0.75 mmol) at 0 °C under N2. The resultant mixture was stirred at room temperature. The bright yellow color faded slowly and, finally, a control TLC proved that the starting material had been consumed (3–4 h). The reaction was quenched by addition of a saturated solution of $NH₄Cl$ (5 mL) and the resulting suspension was diluted with CH_2Cl_2 (15 mL). The organic layer was washed with a 0.5 M NaOH aqueous solution (5 mL), and water (5 mL), and dried (Na₂SO₄). Removal of the volatiles afforded a colorless oil, which was purified through a short pad of silica when required. Yield: 95-100%; colorless oil; R_f 0.30 (CH₂Cl₂); [α]_D –73.0 (c 1.5, CHCl₃); IR (film): 2958, 1740, 1459, 1268, 1167, 1098; ¹H NMR (CDCl₃, 300 MHz): δ 7.35-7.30 (5H, m, ArH), 4.25 (1H, d, J=9.8, CHOCH₃), 3.76 (3H, s, COOCH₃), 3.15 (3H, s, CHOCH₃), 2.77 (1H, dq, J=9.8, 7.0, COCHCH₃), 0.87 (1H, d, J=7.0, COCHCH₃); ¹³C NMR (CDCl₃, 75.4 MHz): δ 175.8, 138.8, 128.4, 128.2, 127.6, 85.8, 56.7, 51.7, 46.9, 14.0; MS–CI (NH3): m/z (%) 209 (35) $[M+H]^{+}$, 226 (100) $[M+NH_4]^{+}$; HRMS (+ESI): calcd for $[M-OMe]^{+}$ $C_{11}H_{13}O_2$ 177.0910, found 177.0904; calcd for $[M+H]^+ C_{12}H_{17}O_3$ 209.1172, found 209.1167.

3.7. Preparation of (S)-4-isopropyl-1,3-oxazolidine-2 thione (3e)

Anhydrous triethylamine (5.4 mL, 38.7 mmol) was added dropwise to a solution of (S)-valinol (1.01 g, 9.8 mmol) and carbon disulfide (3.0 mL, 50.0 mmol) in anhydrous THF (40 mL) at 0° C under N_2 and the resultant mixture was heated at reflux for 12 h under N_2 . After cooling, the volatiles were removed with a rotary evaporator and the mixture was partitioned with CH_2Cl_2 (150 mL) and water (50 mL). The aqueous layer was further extracted with CH_2Cl_2 (3×100 mL), the combined organic layers were dried (Na2SO4), filtered, and concentrated. The resultant brown residue was purified through a column chromatography (CH_2Cl_2) to afford 1.07 g (7.4 mmol, 75% yield) of (S)-4-isopropyl-1,3-oxazolidine-2- thione (3e). White solid; mp 49–50 °C [lit.^{[12a](#page-6-0)} mp 45–46 °C]; R_f 0.20 (CH₂Cl₂); [α]_D –27.4 (c 0.65, CHCl₃) [lit.^{[12a](#page-6-0)} [α]_D –21.4 (c 0.4, CHCl₃)]; IR (KBr): 3191, 2964, 1526, 1274, 1171; ¹H NMR (CDCl₃, 300 MHz): δ 8.46 (1H, br s, NH), 4.69 (1H, t, J=9.1, OCH_xH_V), 4.39 (1H, dd, J=9.1, 6.7, OCH_xH_y), 3.86 (1H, dt, J=9.1, 6.7, NCH), 1.84 (1H, oct, J=6.7, $CH(CH₃)₂$), 0.99 (3H, d, J=6.7, CH₃), 0.94 (3H, d, J=6.7, CH₃); ¹³C NMR (CDCl3, 75.4 MHz): d 189.5, 73.4, 62.4, 32.1, 17.9, 17.8; MS–CI (NH3): m/z (%) 146 (100) $[M+H]^{+}$.

3.8. Preparation of (S)-4-tert-butyl-1,3-oxazolidine-2 thione (3f)

A 2.5 M KOH in 1:1 EtOH/H2O solution (15 mL) was added dropwise to a solution of (S)-tert-leucinol (2.52 g, 21.5 mmol) and carbon disulfide (3.9 mL, 65 mmol) in ethanol (5 mL) at room temperature under N_2 and the reaction mixture was stirred and heated at reflux for 72 h under N_2 . After cooling, the volatiles were removed with a rotary evaporator. Then, a 2 M aqueous HCl (350 mL) was carefully added at room temperature until pH 2 and the resulting mixture was extracted with CH_2Cl_2 (3×50 mL). The combined organic layers were dried ($MgSO₄$), filtered, and the solvent was removed. The resulting white solid was purified by flash chromatography (CH₂Cl₂) to afford 0.32 g (1.8 mmol, 8%) of **1f** and 2.46 g (15.4 mmol, 72%) of (S)-4-tert-butyl-1,3-oxazolidine-2- thione (3f). White solid; mp 146–149 °C [lit.^{[24](#page-7-0)} mp 153–156 °C]; R_f 0.35 (CH2Cl2); [α]_D –8.4 (c 1.0, CHCl3) [lit.²⁴ [α]_D –11.8 (c 0.98, CHCl₃)]; IR (KBr): 3184, 2960, 1534, 1285, 1183; ¹H NMR (CDCl₃, 300 MHz): δ 8.40 (1H, br s, NH), 4.62 (1H, t, J=9.6, OCH_xH_v), 4.46 $(1H, dd, J=9.6, 6.3, OCH_xH_v), 3.81 (1H, dd, J=9.6, 6.3, NCH), 0.94 (9H,$ s, C(CH₃)₃); ¹³C NMR (CDCl₃, 75.4 MHz) δ 189.6, 71.8, 65.8, 33.6,

3.9. Acylation of the 1,3-oxazolidine-2-thiones

25.0; MS–CI (NH₃): m/z (%) 160 (100) [M+H]⁺.

The acylation of oxazolidinethiones 3 was carried out following the experimental procedure reported for the thiazolidinethiones 1.

3.9.1. (S)-4-Isopropyl-N-propanoyl-1,3-oxazolidine-2-thione (7e)

Yield: 85%; white solid; mp 42-43 °C [lit.^{[25](#page-7-0)} mp 43-44 °C]; R_f 0.65 (CH₂Cl₂); $[\alpha]_D + 126.6$ (c 1.05, CHCl₃); $[\text{lit.}^{25} [\alpha]_D + 140.2$ $[\text{lit.}^{25} [\alpha]_D + 140.2$ $[\text{lit.}^{25} [\alpha]_D + 140.2$ (c 1.07, CHCl₃)]; IR (KBr): 2968, 1706, 1465, 1405, 1328, 1200; ¹H NMR (CDCl3, 300 MHz): d 4.79–4.64 (1H, m, NCH), 4.42–4.38 (2H, m, OCH₂), 3.41 (1H, dq, J=18.1, 7.2, COCH_xH_v), 3.27 (1H, dq, J=18.1, 7.2, COCH_xH_y), 2.36 (1H, dhept, J=6.9, 3.9, CH(CH₃)₂), 1.20 (3H, t, J=7.2, COCH₂CH₃), 0.95 (3H, d, J=6.9, CH₃CHCH₃), 0.90 (3H, d, J=6.9, CH₃CHCH₃); ¹³C NMR (CDCl₃, 75.4 MHz): δ 186.0, 174.8, 67.5, 63.2, 31.3, 28.9, 18.2, 14.9, 8.5; HRMS (+ESI): calcd for $[M+H]$ ⁺ $C_9H_{16}NO_2S$ 202.0896, found 202.0899.

3.9.2. (S)-4-tert-Butyl-N-propanoyl-1,3-oxazolidine-2-thione (7f)

Yield: 95%; colorless oil; R_f 0.40 (50:50 hexanes/CH₂Cl₂); $[\alpha]_D$ $+152.2$ (c 1.1, CHCl₃); IR (film): 2968, 1708, 1402, 1326, 1179, 1015; ¹H NMR (CDCl₃, 300 MHz): δ 4.78 (1H, dd, J=7.5, 1.7, NCH), 4.45 (1H, dd, J=9.5, 1.7, OCH_xH_v), 4.34 (1H, dd, J=9.5, 7.5, OCH_xH_v), 3.32 (1H, dq, J=18.1, 7.2, COCH_xH_v), 3.25 (1H, dq, J=18.1, 7.2, COCH_xH_v), 1.21 (3H, t, J=7.2, COCH₂CH₃), 0.94 (9H, s, C(CH₃)₃); ¹³C NMR (CDCl3, 75.4 MHz): d 187.0, 174.9, 69.2, 65.1, 36.1, 31.1, 25.8, 8.9; HRMS (+ESI): calcd for $[M+H]^+$ C₁₀H₁₈NO₂S 216.1052, found 216.1053.

3.10. General procedure for the titanium-mediated addition of 7 to PhCH(OMe)₂

The cross-coupling reaction of N-propanoyl-1,3-oxazolidine-2-thiones 7 and PhCH(OMe)₂ was carried out according to the experimental procedure described for 2.

3.10.1. (S)-4-Isopropyl-N-[(2R,3S)-3-methoxy-2-methyl-3 phenylpropanoyl]-1,3-oxazolidine-2-thione (**9e**)

Yield of mixture: 57%; colorless viscous oil; R_f 0.45 (CH₂Cl₂); HPLC (98:2 hexanes/EtOAc) t_R 40.2 min [2,3-syn diastereomer, t_R 48.7 min]; ¹H NMR (CDCl₃, 300 MHz): δ 7.40–7.25 (5H, m, ArH), 5.35 (1H, dq, J=9.9, 6.9, COCHCH₃), 4.86 (1H, ddd, J=7.7, 4.9, 3.8, NCH), 4.45–4.35 (2H, m, OCH₂), 4.33 (3H, d, J=9.9, CHOCH₃), 3.08 (3H, s, OCH₃), 2.37 (1H, heptd, J=7.0, 3.8, CH(CH₃)₂), 0.97 (3H, d, J=7.0, CH₃), 0.96 (3H, d, J=7.0, CH₃), 0.90 (3H, d, J=6.9, CH₃); ¹³C NMR (CDCl₃, 75.4 MHz) δ 185.5, 175.2, 139.2, 128.2, 127.9, 127.8, 84.3, 66.9, 62.5, 56.8, 44.6, 28.6, 18.1, 14.4, 13.2.

3.10.2. (S)-4-tert-Butyl-N-[(2R,3S)-3-methoxy-2-methyl-3 phenylpropanoyl]-1,3-oxazolidine-2-thione (9f)

Yield: 56%; colorless oil; R_f 0.75 (CH₂Cl₂); HPLC (96:4 hexanes/ EtOAc) t_R 26.3 min [2,3-syn diastereomer, t_R 30.8 min]; [α]_D +34.4 (c 0.4 in CHCl₃); IR (film): 2965, 1701, 1456, 1358, 1254, 1174; ¹H NMR (CDCl₃, 300 MHz): δ 7.42-7.29 (5H, m, ArH), 5.46 (1H, dq, $J=10.0$, 6.9, COCHCH₃), 4.91 (1H, dd, J=7.8, 1.8, NCH), 4.45 (1H, dd, J=9.5, 1.8, OCH_xH_v), 4.36 (3H, d, J=10.0, CHOCH₃), 4.32 (1H, dd, J=9.5, 7.8, OCH_xH_y), 3.08 (3H, s, OCH₃), 1.01 (9H, s, C(CH₃)₃), 0.84 (3H, d, J=6.9, COCHCH₃); ¹³C NMR (CDCl₃, 75.4 MHz): δ 186.9, 177.4, 139.1, 128.5, 128.3, 128.2, 87.4, 68.6, 65.0, 56.4, 43.8, 36.2, 25.6, 14.3; HRMS (+ESI): calcd for $[M+H]^+$ C₁₈H₂₆NO₃S 336.1627, found 336.1616; calcd for $[M-OMe]^+$ $C_{17}H_{22}NO_2S$ 304.1365, found 304.1355.

Acknowledgements

Financial support from the Spanish Ministerio de Ciencia y Tecnología and Fondos FEDER (Grants BQU2002-1514 and CTQ2006-13249/BQU), and the Generalitat de Catalunya (2005SGR00584), as well as doctorate studentship to A.C. (Generalitat de Catalunya) and E.G. (Universitat de Barcelona) are acknowledged.

References and notes

- 1. Nicolaou, K. C.; Sorensen, E. J. Classics in Total Synthesis; VCH: Weinheim, 1996. 2. For recent contributions, see: (a) Gnas, Y.; Glorius, F. Synthesis 2006, 1899; (b) Evans, D. A.; Helmchen, G.; Rüping, M.; Brückner, R.; Davis, F. A.; Enders, D.; Bettray, W.; Hoffmann, R. W.; Kunz, H.; Meyers, A. I.; Murahashi, S.-I.; Imada, Y. Asymmetric Synthesis-The Essentials; Christmann, M., Bräse, S., Eds.; Wiley-VCH: Weinheim, 2008; Part I, pp 3–47.
- 3. Seyden-Penne, J. Chiral Auxiliaries and Ligands in Asymmetric Synthesis; John Wiley & Sons: New York, NY, 1995.
- 4. Evans, D. A.; Bartroli, J.; Shih, T. L. J. Am. Chem. Soc. 1981, 103, 2127.
- 5. (a) Ager, D. J.; Prakash, I.; Schaad, D. R. Aldrichimica Acta 1997, 30, 3; (b) Evans, D. A.; Shaw, J. T. Actual. Chim. 2003, 35.
- 6. Nerz-Stormes, M.; Thornton, E. R. J. Org. Chem. 1991, 56, 2489.
- 7. Nagao, Y.; Hagiwara, Y.; Kumagai, T.; Ochiai, M.; Inoue, T.; Hashimoto, K.; Fujita, E. J. Org. Chem. 1986, 51, 2391.
- 8. (a) González, A.; Aiguadé, J.; Urpí, F.; Vilarrasa, J. Tetrahedron Lett. **1996**, 37, 8949; (b) Zhang, Y.; Phillips, A. J.; Sammakia, T. Org. Lett. **2004**, 6, 23; (c) Zhang Y.; Sammakia, T. Org. Lett. 2004, 6, 3139; (d) Crimmins, M. T.; Shamszad, M. Org. Lett. 2007, 9, 149; (e) Osorio-Lozada, A.; Olivo, H. F. Org. Lett. 2008, 10, 617.
- 9. For similar processes based on N-acetyl-1,3-oxazolidine-2-thiones, see: (a) Yan, T.-H.; Hung, A.-W.; Lee, H.-C.; Chang, C.-S.; Liu, W.-H. *J. Org. Chem.* **1995**, 60,
3301; (b) Guz, N. R.; Phillips, A. J. *Org. Lett.* **2002**, 4, 2253.
- 10. For recent reviews on the synthesis and reactivity of chiral oxazolidinethiones and thiazolidinethiones, see: (a) Velázquez, F.; Olivo, H. F. Curr. Org. Chem. 2002, 6, 1; (b) Ortiz, A.; Sansinenea, E. J. Sulfur Chem. 2007, 29, 109.
- 11. (a) Cosp, A.; Romea, P.; Talavera, P.; Urpí, F.; Vilarrasa, J.; Font-Bardia, M.; Solans, X. Org. Lett. 2001, 3, 615; (b) Cosp, A.; Romea, P.; Urpí, F.; Vilarrasa, J. Tetrahedron Lett. 2001, 42, 4629.
- 12. For the preparation of thiazolidinethione-like chiral auxiliaries, see: (a) Delaunay, D.; Toupet, L.; Le Corre, M. J. Org. Chem. 1995, 60, 6604; (b) See Ref. 10.
- 13. Other Lewis acids (BCl₃, TiCl₄, Et₂AlCl, TMSOTf, Ti(i-PrO)₄, MgCl₂·OEt₂, ZnCl₂, and $LaCl₃$) were also surveyed, but all of them turned out to be less suitable. Otherwise, the same overall yield (87%) and diastereomeric ratio (dr 88:12) were achieved with $SnCl₄$, but $BF₃·OEt₂$ was chosen because its milder character as Lewis acid.
- 14. Crystallographic data (excluding structure factors) for this structure has been deposited with the Cambridge Crystallographic Data Centre as supplementary publication number CCDC-150269. Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: þ44 (0) 1223 336033 or e-mail: deposit@ccdc.cam.ac.uk).
- 15. For a structural study on these adducts, see: Cosp, A.; Larrosa, I.; Anglada, J. M.; Bofill, J. M.; Romea, P.; Urpí, F. Org. Lett. 2003, 5, 2809.
- 16. The steric hindrance of the phenyl group has not been clearly established. According to the A values (for a comprehensive list, see: Eliel, E. L.; Wilen, S. H. Stereochemistry of Organic Compounds; John Wiley & Sons: New York, NY, 1994), it is more bulky $(A=2.8 \text{ kcal mol}^{-1})$ than other groups such as isopropyl $(A=2.21 \text{ kcal mol}^{-1})$. However, it is accepted that it becomes smaller than a methyl group $(A=1.74 \text{ kcal mol}^{-1})$ on the basis of the different conformational arrangements available for the alternative transition states (for a nice discussion, see: Roush, W. R. J. Org. Chem. 1991, 56, 4151). Since the diastereoselectivity of the cross-coupling addition of the titanium enolates from 3 to PhCH(OMe)₂ relies on kinetic grounds, it may be helpful to consider the phenyl group as the smallest substituent positioned at C4 in the N-propanoyl thiazolidinethiones 3.
- 17. For the effect of the C4 substituent of such chiral auxiliaries in the context of the acetate aldol reaction, see Ref. [8b.](#page-6-0)
- 18. An excellent analysis on the steric bulk of the C4 substituent in the addition of the titanium enolates from N-acetyl-1,3-thiazolidine-2-thiones to dimethyl acetals in the course of the total synthesis of $(-)$ -hennoxazole has been recently reported, see: Smith, T. E.; Kuo, W.-H.; Balskus, E. P.; Bock, V. D.; Roizen, J. L.; Theberge, A. B.; Carroll, K. A.; Kurihara, T.; Wessler, J. D. J. Org. Chem. 2008, 73, 142.
- 19. Lewis acid-mediated additions of carbon nucleophiles to acetals usually proceed through carbocation intermediates. For example, see: (a) Sammakia, T.; Smith, R. S. J. Am. Chem. Soc. 1992, 114, 10998; (b) Sammakia, T.; Smith, R. S. J. Am. Chem. Soc. 1994, 116, 7915; (c) Matsutani, H.; Ichikawa, S.; Yaruva, J.; Kusumoto, T.; Hiyama, T. J. Am. Chem. Soc. 1997, 119, 4541; (d) Romero, J. A. C.; Tabacco, S. A.; Woerpel, K. A. J. Am. Chem. Soc. 2000, 122, 168. For earlier mechanistic discussions, see: (e) Mori, I.; Ishihara, K.; Flippin, L. A.; Nozaki, K.;
Yamamoto, H.; Bartlett, P. A.; Heathcock, C. H. J. Org. *Chem.* **1990**, 55, 6107 and references therein.
- 20. Unpublished results from Erik Galvez, Ph.D. Thesis in course.
- 21. However, it has been also reported that Lewis acid-mediated additions to chiral dioxane acetals can proceed by an S_N2 mechanism. For a nice account on this issue, see: Denmark, S. E.; Almstead, N. G. J. Am. Chem. Soc. 1991, 113, 8089.
- 22. For related examples on the influence of C4 substituents on the structure of several enolates, see: (a) Hintermann, T.; Seebach, D. Helv. Chim. Acta 1998, 81, 2093; (b) Bull, S. D.; Davies, S. G.; Key, M.-S.; Nicholson, R. L.; Savory, E. D. Chem. Commun. 2000, 1721.
- 23. For the crucial role played by the exocyclic sulfur atom in the addition of titanium enolates from N-acetyl-1,3-oxazolidine-2-thiones to 2,5-disubstituted
tetrahydrofurans, see: Jalce, G.; Seck, M.; Franck, X.; Hocquemiller, R.; Figadère, B. J. Org. Chem. 2004, 69, 3240.
-
- 24. Yamada, S.; Katsumata, H. J. Org. Chem. **1999**, 64, 9365.
25. Cuzzupe, A. N.; Hutton, C. A.; Lilly, M. J.; Mann, R. K.; McRae, K. J.; Zammit, S. C.;
Rizzacasa, M. A. J. Org. Chem. **2001**, 66, 2382.